

8th Grade Science Chapter 12 Answers

1. The periodic table was created by a Russian chemist, Dmitri Mendeleev in the 1860's. P 302
2. The term periodic means, "to have a regular or repeating pattern."
3. The elements are arranged in a Periodic Table because the properties of these elements change in a periodic (regular or repeating pattern) way. Think about the calendar, every week has 7 days, it is a "period" of time.
4. Mendeleev arranged his elements in order of their increasing atomic mass. P 303
5. Mosley helped to correct Mendeleev's order by atomic mass. Instead of using atomic mass, Mosley determined that it was best to use the elements atomic number or the number of protons in the atom. Once Mosley was done, all of the elements fell into their proper place on the table.
6. The Periodic Law is a principle that states that the chemical and physical properties of the elements are periodic functions of their atomic numbers.
7. the 6 = atomic #
 C = symbol
 Carbon = Element Name
 12.0 = atomic mass
8. The background colors of the Periodic Table actually indicate what type of element it is. The Blue background is for a metals, the Green's are metalloids, and the yellow indicates a nonmetal.
9. The color of the chemical symbol indicates the physical state of the element at room temperature. If the symbol is: Red = Solid Blue = Liquid Green = Gas
10. A row of elements is called a Period.
11. A column of elements is called a Group or Family
12. The number in the parenthesis is the mass number of the most stable isotope of the element.
13. Yes, the elements of the symbols are universally recognized by an International Committee of scientists that approves the name and symbol for each element.
14. Elements are classified as a metal, nonmetal, or a metalloid, according to their properties.
15. The number of electrons in the outer energy level is used to determine which category an element is placed.
16. Four properties of metals are: Good Conductors of Thermal Energy, Malleable: they can be flattened with a hammer without shattering, Ductile: they can be drawn into thin wires and are then good conductors of electric current, Shiny: you can see a reflection.
17. Nonmetals are: Poor Conductors of Thermal Energy: can't conduct thermal energy, Not Malleable: can't be flattened with a hammer without shattering, Not Ductile: can not be drawn into wires, & Not Shiny.

8th Grade Science Chapter Review Answers

Ying-Ying Zheng



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Table of Contents 8th Grade Science Chapter Review Answers

1. Understanding the eBook 8th Grade Science Chapter Review Answers
 - The Rise of Digital Reading 8th Grade Science Chapter Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying 8th Grade Science Chapter Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 8th Grade Science Chapter Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from 8th Grade Science Chapter Review Answers
 - Personalized Recommendations
 - 8th Grade Science Chapter Review Answers User Reviews and Ratings
 - 8th Grade Science Chapter Review Answers and Bestseller Lists

5. Accessing 8th Grade Science Chapter Review Answers Free and Paid eBooks
 - 8th Grade Science Chapter Review Answers Public Domain eBooks
 - 8th Grade Science Chapter Review Answers eBook Subscription Services
 - 8th Grade Science Chapter Review Answers Budget-Friendly Options
6. Navigating 8th Grade Science Chapter Review Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - 8th Grade Science Chapter Review Answers Compatibility with Devices
 - 8th Grade Science Chapter Review Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 8th Grade Science Chapter Review Answers
 - Highlighting and Note-Taking 8th Grade Science Chapter Review Answers
 - Interactive Elements 8th Grade Science Chapter Review Answers
8. Staying Engaged with 8th Grade Science Chapter Review Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 8th Grade Science Chapter Review Answers
9. Balancing eBooks and Physical Books 8th Grade Science Chapter Review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 8th Grade Science Chapter Review Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 8th Grade Science Chapter Review Answers
 - Setting Reading Goals 8th Grade Science Chapter Review Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 8th Grade Science Chapter Review Answers
 - Fact-Checking eBook Content of 8th Grade Science Chapter Review Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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