

## Section 9.2: Team Learning Worksheet

1. Consider the following unbalanced equation:



You have 5.00 grams of manganese(IV) oxide:

- a. How many moles of manganese(IV) oxide do you have?
  
  
  
  
  
  
  
  
  
  
- b. How many moles of HCl do you need for a complete reaction?
  
  
  
  
  
  
  
  
  
  
- c. How many grams of HCl do you need for a complete reaction?
  
  
  
  
  
  
  
  
  
  
- d. How many grams of each product ( $\text{H}_2\text{O}$ ,  $\text{MnCl}_2$ , and  $\text{Cl}_2$ ) would be formed?
  
  
  
  
  
  
  
  
  
  
- e. Prove that mass has been conserved in this reaction.
  
  
  
  
  
  
  
  
  
  
- f. Is the number of atoms conserved? Support your answer.
  
  
  
  
  
  
  
  
  
  
- g. Is the number of molecules conserved? Support your answer.

# 9 1 Team Learning Worksheet

**David Kirk**



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