

Name: \_\_\_\_\_

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Chemistry 11

**Classifying Matter**

Assignment

1. Complete the following sentences by filling in the appropriate word from the list below.

Elements  
Mixtures  
Compound  
Symbols  
Properties

Carbon  
Pure  
Letters  
Carbon dioxide

- a. \_\_\_\_\_ **Elements** \_\_\_\_\_ cannot be separated by physical or chemical means.
- b. Elements are abbreviated with \_\_\_\_\_ **symbols** \_\_\_\_\_, which consist of one or two \_\_\_\_\_ **letters** \_\_\_\_\_.
- c. Two or more elements chemically combined make up a \_\_\_\_\_ **compound** \_\_\_\_\_.
- d. \_\_\_\_\_ **Carbon** \_\_\_\_\_ is an example of an element and \_\_\_\_\_ **carbon dioxide** \_\_\_\_\_ is an example of a compound.
- e. Elements and compounds are called \_\_\_\_\_ **pure** \_\_\_\_\_ substances because they have a unique set of chemical and physical \_\_\_\_\_ **properties** \_\_\_\_\_.

2. Fill in the blanks in the following chart.

Element Name	Element Symbol	Common Ions Formed
a. Sodium	<b>Na</b>	<b>+1</b>
b. Copper	Cu	<b>+2, +1</b>
c. Tungsten	W	<b>+6</b>
d. Iron	<b>Fe</b>	<b>+3, +2</b>
e. Tin	Sn	<b>+4, +2</b>
f. Potassium	<b>K</b>	<b>+1</b>
g. Gold	Au	<b>+3, +1</b>
h. Silver	Ag	<b>+1</b>

3. Classify each of the following as one of an Atom (A), a Molecule (M) or an Ion (I).

a.  $P^{3-}$       I  

b.  $H_2O$       M  

c.  $O_2$       M  

d.  $Au^{3+}$       I  

e. Ge      A  

f. O      A  

g.  $Ca^{2+}$       I  

h.  $NH_3$       M

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**Herbert Drapkin**



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