
AP CHEMISTRY FINAL EXAM REVIEW 2

1. Is energy absorbed or released in an endothermic reaction?: Absorbed.
2. What is the sign of delta H for an exothermic reaction?: Exothermic reactions release energy. The sign of delta H is negative.
3. When you touch a flask and it feels hot, is this an endothermic or exothermic process?: Exothermic; heat is being given off.
4. Define specific heat capacity.: The amount of heat required to raise 1g of a substance to 1 degrees Celsius.
5. $q=mc(\Delta T)$; What does "q" stand for?: Heat (kJ)
6. $q=mc(\Delta T)$; What does "c" stand for?: Specific heat capacity (J / g degrees C)
7. What is a hydrogen ion donor called?: An acid.
8. What is a hydrogen ion acceptor called?: A base.
9. Feels slippery; acid or base?: Base.
10. Tastes sour; acid or base?: Acid.
11. Tastes bitter; acid or base?: Base.
12. $\text{pH} < 7$; acid, base, or neutral?: Acid
13. $\text{pH} > 7$; acid, base, or neutral?: Base
14. Define neutralization.: When an acid reacts with a base to form water and a salt, $\text{pH} = 7$.
15. Soap; acid or base?: Base (tastes bitter)
16. Ammonia, acid or base?: Base (drain cleaners are bases).
17. Lemonade; acid or base?: Acid (tastes sour)
18. How many valence electrons are in an atom of chlorine?: 7
19. How many valence electrons does aluminum have?: 3
20. Define Reaction Rate: The decrease in concentration of reactants or the increase in concentration of products with time.
21. How does increasing temperature affect the rate of reaction?: Increasing temperature causes the kinetic energy to increase. Thus, molecules are moving faster and colliding more frequently. This causes the rate of reaction to increase.
22. If I decrease the concentration of reactant, what happens to the rate of reaction?: Decreasing concentration leads to less reactants available for collision. Therefore, reaction rate decreases.
23. How does surface area influence reaction rate?: By increasing the surface area of reactant molecules, we are increasing the space or point of contact that these molecules have available

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M Mark



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Bobrow Test Preparation Services,2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter s end Two full length practice tests with detailed answer explanations are included in the book **Scientific and Technical Aerospace Reports** ,1969 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database Cracking the AP Chemistry Exam, 2016 Edition Princeton Review,2015-08-11 EVERYTHING YOU NEED TO HELP SCORE A PERFECT 5 Equip yourself to ace the AP Chemistry Exam with The Princeton Review s comprehensive study guide including 2 full length practice tests thorough content reviews access to our AP Connect online portal and targeted strategies for every section of the exam This eBook edition is optimized for on screen learning with cross linked questions answers and explanations We don t have to tell you how tough AP Chem is or how important a stellar score on the AP exam can be to your chances of getting into a top college of your choice Written by Princeton Review experts who know their way around chem Cracking the AP Chemistry Exam will give you the help you need to get the score you want Techniques That Actually Work Tried and true strategies to help you avoid traps and beat the test Tips for pacing yourself and guessing logically Essential tactics to help you work smarter not harder Everything You Need to Know to Help Achieve a High Score Comprehensive content review for all test topics Up to date information on the 2016 AP Chemistry Exam Engaging activities to help you critically assess your progress Access to AP Connect our online portal for helpful pre college information and exam updates Practice Your Way to Excellence 2 full length practice tests with detailed answer explanations Practice drills at the end of each content chapter Review of important laboratory procedures and equipment *Synerjy* ,1983 **Resources in Education** ,1976 **Pacific Rural Press** ,1899 **EPA Publications Bibliography** ,1994 **Normal Instructor and Primary Plans** ,1918 **Government Reports Announcements & Index** ,1993-05 **The Mining World** ,1907 **Mining and Engineering World** ,1906 **Resources in Education** ,1997 The Mining Investor ,1912 *Educational Times* ,1891 *Financial World* ,1923 **Energy Research Abstracts** ,1982 **Mining American** ,1915 **American Orchid Society Bulletin** American Orchid Society,1981

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