

Stoichiometry

Section 11.1 What is stoichiometry?

In your textbook, read about stoichiometry and the balanced equation.

For each statement below, write *true* or *false*.

- _____ 1. The study of the quantitative relationships between the amounts of reactants used and the amounts of products formed by a chemical reaction is called stoichiometry.
- _____ 2. Stoichiometry is based on the law of conservation of mass.
- _____ 3. In any chemical reaction, the mass of the products is less than the mass of the reactants.
- _____ 4. The coefficients in a chemical equation represent not only the number of individual particles but also the number of moles of particles.
- _____ 5. The mass of each reactant and product is related to its coefficient in the balanced chemical equation for the reaction by its molar mass.

Complete the table below, using information represented in the chemical equation for the combustion of methanol, an alcohol.

methanol + oxygen \rightarrow carbon dioxide + water



| Substance | Molar Mass (g/mol) | Number of Molecules | Number of Moles (mol) | Mass (g) |
|-------------------|--------------------|---------------------|-----------------------|----------|
| 6. Methanol | 32.05 | | | |
| 7. Oxygen gas | 32.00 | | | |
| 8. Carbon dioxide | 44.01 | | | |
| 9. Water | 18.02 | | | |

10. What are the reactants? _____
11. What are the products? _____
12. What is the total mass of the reactants? _____
13. What is the total mass of the products? _____
14. How do the total masses of the reactants and products compare? _____

Chapter 12 Study Guide For Content Mastery Stoichiometry Answers

RS Peters



Chapter 12 Study Guide For Content Mastery Stoichiometry Answers:

Enjoying the Beat of Term: An Mental Symphony within **Chapter 12 Study Guide For Content Mastery Stoichiometry Answers**

In a global taken by screens and the ceaseless chatter of immediate interaction, the melodic beauty and mental symphony developed by the published word frequently diminish into the back ground, eclipsed by the constant sound and interruptions that permeate our lives. Nevertheless, set within the pages of **Chapter 12 Study Guide For Content Mastery Stoichiometry Answers** an enchanting literary value overflowing with natural thoughts, lies an immersive symphony waiting to be embraced. Constructed by an outstanding composer of language, that charming masterpiece conducts readers on a mental journey, skillfully unraveling the concealed melodies and profound influence resonating within each carefully constructed phrase. Within the depths with this emotional evaluation, we can discover the book is key harmonies, analyze its enthralling publishing fashion, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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