

## Chapter 15 – Acids and Bases

- Acids – have sour taste
  - Vinegar – acetic acid
  - Citrus fruits – citric acid
  - React w metals to produce hydrogen gas
  - React with carbonates and bicarbonates to produce carbon dioxide gas
  - Change the color of certain dyes – green, yellow, blue
- Bases
  - Taste bitter
  - Feel slippery – some soaps
- Bronsted-Lowry
  - Bronsted acid - proton donor
  - Bronsted base – proton acceptor
- Arrhenius theory
  - Acid – anything that produces  $H_3O^+$  in aqueous solution
  - Base – anything that produces  $OH^-$  in aqueous solutions
- When an acid donates proton  $\rightarrow$  conjugate base
- When a base accepts a proton  $\rightarrow$  conjugate acid
- The ion-product constant ( $K_w$ ) is the product of the molar concentrations of  $H^+$  and  $OH^-$  ions at a particular temperature.
- Conjugate acid-base pairs
  - The conjugate base of a strong acid has no measurable strength.
  - $H_3O^+$  is the strongest acid that can exist in aqueous solution.
  - The  $OH^-$  ion is the strongest base that can exist in aqueous solution.
- SLIDE 27 – MEMORIZE
- Diprotic and triprotic acids
  - Diprotic- 2 protons to donate
  - Triprotic- 3 protons to donate
  - May yield more than one hydrogen ion per molecule
  - Ionize in a stepwise manner; they lose one proton at a time
  - An ionization constant expression can be written for each ionization stage
  - Two or more equilibrium constant expressions must often be used to calculate the concentrations of species in the acid solution
- Binary acids –  $(H)_\text{_____}$ 
  - Ex. HF
- Oxoacids –  $(H)_\text{_____}(O)_\text{_____}$ 
  - Ex. HNO<sub>3</sub>
  - \_\_\_\_\_: any element
  - \_\_\_\_\_

# Chapter 15 Acids Bases Study Guide

**Sabine Zange**



## **Chapter 15 Acids Bases Study Guide:**

## The Enigmatic Realm of **Chapter 15 Acids Bases Study Guide**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Chapter 15 Acids Bases Study Guide** a literary masterpiece penned with a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of people who partake in its reading experience.

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