Worksheet: Boyle's Law and Charles's Law **Name** KEY

1. Boyle's Law: When temperature is held constant, the pressure and volume of a gas are inversely proportional.

2. Mathematically, Boyle's law is stated PV = k or $P_1V_1 = P_2V_2$.

3. At a pressure of 405 kPa, the volume of a gas is 6.00 cm^3. Assuming the temperature remains constant, at what pressure will the new volume be 4.00 cm^3?

$P_1 = 405\ kPa$ $P_2 = ?$
$V_1 = 6.00\ cm^3$ $V_2 = 4.00\ cm^3$

$$P_1V_1 = P_2V_2$$

$$405\ kPa \times 6.00\ cm^3 = P_2 \times 4.00\ cm^3 \qquad P_2 = \boxed{608\ kPa}$$

4. A volume of gas at 1.10 atm was measured at 326 cm^3. What will be the volume if the pressure is adjusted to 1.90 atm?

$P_1 = 1.10\ atm$ $P_2 = 1.90\ atm$
$V_1 = 326\ cm^3$ $V_2 = ?$

$$P_1V_1 = P_2V_2$$

$$1.10\ atm \times 326\ cm^3 = 1.90\ atm \times V_2 \qquad V_2 = \boxed{189\ cm^3}$$

5. If 36.5 m^3 of a gas are collected at a pressure of 755 mm Hg, what volume will the gas occupy if the pressure is changed to 632 mm Hg?

$P_1 = 755\ mm\ Hg$ $P_2 = 632\ mm\ Hg$
$V_1 = 36.5\ m^3$ $V_2 = ?$

$$P_1V_1 = P_2V_2$$

$$755\ mmHg \times 36.5\ m^3 = 632\ mmHg \times V_2 \qquad V_2 = \boxed{43.6\ m^3}$$

Answer Key For Boyles Law

Michael Seilmaier

Answer Key For Boyles Law:

Reviewing **Answer Key For Boyles Law**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Answer Key For Boyles Law**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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