



SPECIFIC HEAT WORKSHEETS

$Q = mc\Delta T$,
where Q = heat energy, m = mass, and
 ΔT = change in temp.
Remember, $\Delta T = (T_{\text{final}} - T_{\text{initial}})$.



Show all work and proper units.

- A 15.75-g piece of iron absorbs 1086.75 joules of heat energy, and its temperature changes from 25°C to 175°C. Calculate the specific heat capacity of iron.
 $c = q / (m \cdot \Delta T)$
 $c = 1086.75 \text{ J} / (15.75 \text{ g} \times 150 \text{ }^\circ\text{C})$
 $= 0.46 \text{ J/g }^\circ\text{C}$
- How many joules of heat are needed to raise the temperature of 10.0 g of aluminum from 22°C to 55°C, if the specific heat of aluminum is 0.90 J/g°C?
 $q = mc\Delta T$
 $q = (10.0 \text{ g})(0.90 \text{ J/g }^\circ\text{C})(33 \text{ }^\circ\text{C})$
 $q = 297 \text{ Joules}$
- Calculate the specific heat capacity of a piece of wood if 1500.0 g of the wood absorbs 67,500 joules of heat, and its temperature changes from 32°C to 57°C.
 $c = q / (m \cdot \Delta T)$
 $c = 67,500 \text{ J} / (1500.0 \text{ g} \times 25 \text{ }^\circ\text{C})$
 $= 1.8 \text{ J/g }^\circ\text{C}$
- 100.0 g of 4.0°C water is heated until its temperature is 37°C. If the specific heat of water is 4.18 J/g°C, calculate the amount of heat energy needed to cause this rise in temperature.
 $q = mc\Delta T$
 $q = (100 \times 4.18 \times 33) \text{ J}$
 $q = 13,794 \text{ J}$
- Calculate the heat required to raise 0.6 Kg of sand from 30°C to 90°C? (Specific Heat of sand = 830 J/Kg°C)
 $q = mc\Delta T$
 $q = 830 \text{ J/Kg }^\circ\text{C} \times 0.6 \text{ Kg} \times 60 \text{ }^\circ\text{C}$
 $q = 29880 \text{ J.}$

Answer Key Heat

Christian Drosten



Answer Key Heat:

The book delves into Answer Key Heat. Answer Key Heat is an essential topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Answer Key Heat, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Answer Key Heat
 - Chapter 2: Essential Elements of Answer Key Heat
 - Chapter 3: Answer Key Heat in Everyday Life
 - Chapter 4: Answer Key Heat in Specific Contexts
 - Chapter 5: Conclusion
2. In chapter 1, this book will provide an overview of Answer Key Heat. The first chapter will explore what Answer Key Heat is, why Answer Key Heat is vital, and how to effectively learn about Answer Key Heat.
3. In chapter 2, this book will delve into the foundational concepts of Answer Key Heat. The second chapter will elucidate the essential principles that need to be understood to grasp Answer Key Heat in its entirety.
4. In chapter 3, this book will examine the practical applications of Answer Key Heat in daily life. This chapter will showcase real-world examples of how Answer Key Heat can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Answer Key Heat in specific contexts. This chapter will explore how Answer Key Heat is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Answer Key Heat. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Answer Key Heat.

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