

Con. 3504-08.

CO-5149

(REVISED COURSE)

(2 Hours)

[Total Marks : 75

- N.B. (1) Question No. 1 is compulsory.
(2) From remaining six questions attempt any four.
(3) All questions carry equal marks.
(4) Figures to the right indicate full marks.

1. Attempt any five of the following : 15
- (a) What is polymerisation? What are the conditions for polymerisation?
- (b) A water sample contains :
(i) $\text{Ca}(\text{HCO}_3)_2 = 32.4 \text{ mg/l}$
(ii) $\text{Mg}(\text{HCO}_3)_2 = 29.2 \text{ mg/l}$
(iii) $\text{CaSO}_4 = 13.5 \text{ mg/l}$
Calculate temporary, permanent and total hardness.
(At. wt. C = 40 Mg = 24 H = 1 Cl = 35.5 O = 16 S = 32)
- (c) What are carbon-nanotubes? Explain different types of carbon-nanotubes.
- (d) What are lubricants? List different functions of lubricants.
- (e) What are fuel cells? List the advantage of fuel cells over conventional power plants.
- (f) What is triple point in phase diagram? Explain it with reference to one component water system phase diagram.
- (g) Distinguish between thermoplastics and thermosetting resins.
2. (a) What do you mean by fabrication of plastics? Explain transfer moulding with the help of a neat diagram. 6
- (b) Explain any two of the following terms :— 6
(i) Phases
(ii) Components
(iii) Degree of freedom.
- (c) 1.55 gram of an oil is saponified with 20 ml of $\frac{N}{10}$ alcoholic potassium hydroxide solution. 3
After refluxing the mixture, it requires 15 ml of $\frac{N}{10}$ HCl solution. Find saponification value of oil.
3. (a) Explain theory, procedure and limitations of Zeolite process with the help of a neat diagram. 6
- (b) Explain the following terms :— 6
(i) Glass transition temperature
(ii) Liquid crystal properties of polymers
(iii) Supramolecules
- (c) State and explain condensed phase rule. 3

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B Lingard



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