Unit 1 study guide

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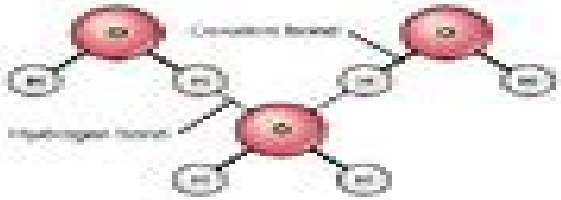
Class	Unit 1 : Chemistry of Life
Reviewed	☐

Topic 1.1 : Structure of Water and Hydrogen Bonding

1. Structure /Polarity

- 2 elements : Oxygen and Hydrogen (1:2 ratio)
- Oxygen is more electronegative than hydrogen , resulting in an unequal sharing of electrons between oxygen and hydrogen. This is called Polarity, where there are differences in atoms electronegativity , a water molecule has polarity
- The bond is a covalent bond between the two elements (they share electrons)
- The bind between other water molecules are hydrogen bond (a weak bond between the negative and positive regions of two separate molecules).



2. Properties of water

LIVING THINGS DEPEND ON WATER'S PROPERTIES

- When two of the same molecules from hydrogen bonds with each other this is called **cohesion** (water and water) .
- When two different molecules form hydrogen bonds with each other this is call **adhesion** (amino acid and water)

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