

Mixtures and Solutions

Section 12.1: Homogeneous and Heterogeneous Mixtures

Use your textbook, class notes, and resources to answer the following questions.

The table below lists mixtures as either

- _____ **None** _____
- _____ **1.** _____ a homogeneous mixture consisting of one or more substances in the same physical state.
- _____ **2.** _____ The two substances uniformly mixed at the molecular level.
- _____ **3.** _____ A mixture of two immiscible liquids.
- _____ **4.** _____ A homogeneous mixture of a dissolved solid in a liquid.
- _____ **5.** _____ Mixture of two or more different elements.
- _____ **6.** _____ An unreacted heterogeneous mixture is formed.
- _____ **7.** _____ A mixture of two or more different particles with different physical and chemical properties.
- _____ **8.** _____ A mixture of two or more different particles with different physical and chemical properties.
- _____ **9.** _____ A mixture of two or more different particles with different physical and chemical properties.
- _____ **10.** _____ A mixture of two or more different particles with different physical and chemical properties.
- _____ **11.** _____ A mixture of two or more different particles with different physical and chemical properties.

Use your textbook, class notes, and resources to answer the following questions. Place a checkmark in the appropriate column for each mixture type.

Mixture Description	Homogeneous	Heterogeneous	None
1. A mixture of two or more different elements.			
2. A mixture of two or more different elements.			
3. A mixture of two or more different elements.			
4. A mixture of two or more different elements.			
5. A mixture of two or more different elements.			
6. A mixture of two or more different elements.			
7. A mixture of two or more different elements.			
8. A mixture of two or more different elements.			
9. A mixture of two or more different elements.			
10. A mixture of two or more different elements.			
11. A mixture of two or more different elements.			

Chapter 1 study Guide Mixtures And Solutions

AN Whitehead



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