

**Organic Chemistry, 8e (Wade)**

Chapter 1 Introduction and Review

1) While you were up late one night studying organic chemistry, you happened to see the last 5 minutes of an infomercial on TV. The spokesperson claimed that their brand of automobile tires were superior to all other brands on the market because they were made by using only natural rubber, isolated from the resin of rubber trees. How could a chemist test her claims that no petroleum products went into the manufacture of her brand of tires?

Answer: Compounds synthesized from petroleum products have a lower content of  $^{14}\text{C}$ . Plant-derived compounds are recently synthesized from  $\text{CO}_2$  in the air and have a higher  $^{14}\text{C}$  content.

Diff: 2

Section: 1.1

2) The atomic number of boron is 5. The correct electronic configuration of boron is:

A)  $1s^22s^3$

B)  $1s^22p^3$

C)  $1s^22s^22p^1$

D)  $2s^22p^3$

E)  $1s^22s^23s^1$

Answer: C

Diff: 1

Section: 1.2

3) How many distinct p orbitals exist in the second electron shell, where  $n = 2$ ?

A) 2

B) 3

C) 4

D) 5

E) 6

Answer: B

Diff: 1

Section: 1.2

4) The \_\_\_\_\_ tells us that each orbital can hold a maximum of 2 electrons.

A) aufbau principle

B) Pauli exclusion principle

C) Hund's rule principle

D) LeChatelier principle

E) uncertainty principle

Answer: B

Diff: 1

Section: 1.2

# Chapter 1organic Chemistry Wade

**Ying-Ying Zheng**



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