

Name \_\_\_\_\_

answer key

Date \_\_\_\_\_

# Stoichiometry: Mass - Mole Worksheet

Based on the following chemical equation, please answer the following questions!



Calculate the amount of moles of NaCl needed to produce 75 g of FeCl<sub>3</sub>.

$$\frac{9 \text{ FeCl}_3}{1 \text{ mole FeCl}_3} \times \frac{1 \text{ mole FeCl}_3}{9 \text{ FeCl}_3} = \frac{\text{mole NaCl}}{\text{mole FeCl}_3}$$

Calculate the amount of Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> in grams produce 215 g Na<sub>2</sub>SO<sub>4</sub>.

$$\frac{9 \text{ Na}_2\text{SO}_4}{1 \text{ mole Na}_2\text{SO}_4} \times \frac{1 \text{ mole Na}_2\text{SO}_4}{9 \text{ Na}_2\text{SO}_4} = \frac{\text{mole Fe}_2(\text{SO}_4)_3}{\text{mole Na}_2\text{SO}_4} \times \frac{9 \text{ Fe}_2(\text{SO}_4)_3}{1 \text{ mole Fe}_2(\text{SO}_4)_3}$$

Calculate the amount in grams of FeCl<sub>3</sub> produced with 95g of Fe (SO<sub>4</sub>)<sub>3</sub>.

$$\frac{9 \text{ Fe}_2(\text{SO}_4)_3}{1 \text{ mole Fe}_2(\text{SO}_4)_3} \times \frac{1 \text{ mole Fe}_2(\text{SO}_4)_3}{9 \text{ Fe}_2(\text{SO}_4)_3} = \frac{\text{mole FeCl}_3}{\text{mole Fe}_2(\text{SO}_4)_3} \times \frac{9 \text{ FeCl}_3}{1 \text{ mole FeCl}_3}$$

Calculate the amount of moles of Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> needed to react with 123.5 g of NaCl

$$\frac{9 \text{ NaCl}}{1 \text{ mole NaCl}} \times \frac{1 \text{ mole NaCl}}{9 \text{ NaCl}} = \frac{\text{mole Fe}_2(\text{SO}_4)_3}{\text{mole NaCl}}$$

Based on the following chemical equation, please answer the following questions!



$$\frac{9 \text{ Ca(HCO}_3)_2}{1 \text{ mole Ca(HCO}_3)_2} \times \frac{1 \text{ mole Ca(HCO}_3)_2}{9 \text{ Ca(HCO}_3)_2} = \frac{\text{mole CaCO}_3}{\text{mole Ca(HCO}_3)_2}$$

# Chapter 1stoichiometry Practice Problems Answer Key

**Chao Zhang**



**Chapter 1 stoichiometry Practice Problems Answer Key:**

## Unveiling the Power of Verbal Beauty: An Mental Sojourn through **Chapter 1stoichiometry Practice Problems Answer Key**

In some sort of inundated with displays and the cacophony of instantaneous conversation, the profound power and emotional resonance of verbal beauty usually disappear into obscurity, eclipsed by the constant barrage of sound and distractions. However, set within the musical pages of **Chapter 1stoichiometry Practice Problems Answer Key**, a fascinating perform of fictional elegance that impulses with fresh thoughts, lies an memorable trip waiting to be embarked upon. Composed with a virtuoso wordsmith, this enchanting opus books viewers on an emotional odyssey, delicately exposing the latent possible and profound affect stuck within the elaborate web of language. Within the heart-wrenching expanse with this evocative evaluation, we will embark upon an introspective exploration of the book is central styles, dissect its captivating publishing fashion, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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