

RADIOACTIVITY: spontaneous disintegration of the nucleus of certain atoms accompanied by the emission (release) of particles and/or energy

~ **TRANSMUTATION:** the change of one element into another due to changes in the nucleus
 * can be caused by radioactive decay (called _____ transmutation)

- Particles that are emitted during radioactive decay

~ alpha particle (α) - written as ${}^4_2\text{He}$; low energy particle; blocked by paper, skin, clothing



~ beta emission (β) - written as ${}^{0}_{-1}\text{e}$ or ${}^{0}_{-1}\beta$; blocked by plastic, wood, light metals



~ positron emission (${}^{0}_{+1}\beta$); also written as ${}^{0}_{+1}\text{e}$



~ gamma radiation (γ) - no mass or charge, just energy; blocked by lead or thick (1+ m) concrete

* can be caused by bombardment by subatomic particles (called _____ transmutation)

- particle accelerator:

~ proton (a hydrogen nucleus)



~ alpha particle (a helium nucleus)



~ only charged particles can be accelerated - neutrons cannot be accelerated in ordinary particle accelerator

BALANCING NUCLEAR EQUATIONS NOTES

- ~ Nuclear equations are different than regular equations because the atomic number and mass number are given along with the symbol for the element.
- ~ To balance a nuclear equation, the sum of the mass numbers of the substances on one side of the equation must equal the sum of the mass numbers of the substances on the other side of the equation. The same is true for the atomic numbers.

~ EXAMPLE: ${}^{242}_{96}\text{Cm} + \underline{\hspace{1cm}} \rightarrow {}^{246}_{98}\text{Cf} + {}^1_0\text{n}$

~ Top # = mass number... Left side = 242, Right side = 246

So, in blank space, top number must be 4

~ Bottom # = atomic number... Left side = 96, Right side = 98

So, in blank space, bottom number must be 2

~ So far the equation is ${}^{242}_{96}\text{Cm} + {}^4_2\underline{\hspace{0.5cm}} \rightarrow {}^{246}_{98}\text{Cf} + {}^1_0\text{n}$

~ Determine the element using the atomic number. He has an atomic number of 2.

~ So, the final equation is ${}^{242}_{96}\text{Cm} + {}^4_2\text{He} \rightarrow {}^{246}_{98}\text{Cf} + {}^1_0\text{n}$

Chapter 1 nuclear Chemistry Answer Key

Raffaella Di Napoli



Chapter 1 nuclear Chemistry Answer Key:

Decoding **Chapter 1nuclear Chemistry Answer Key**: Revealing the Captivating Potential of Verbal Expression

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