

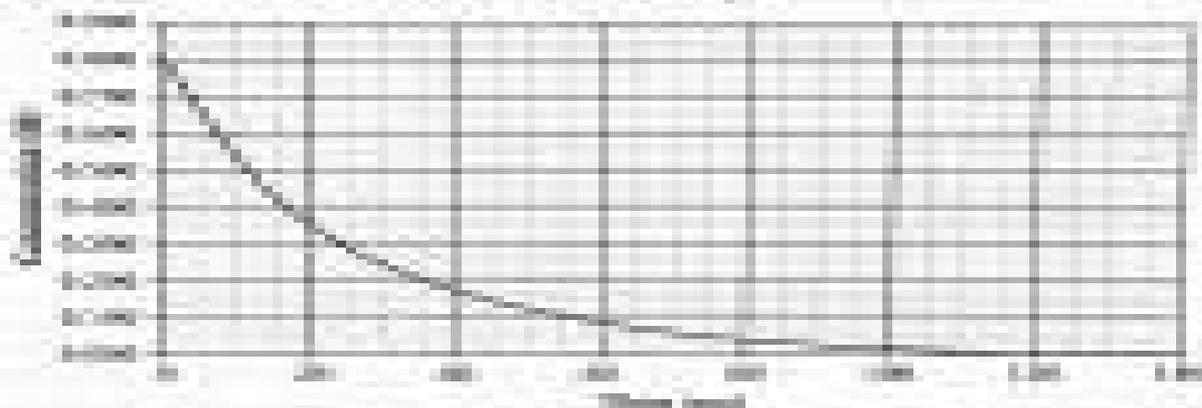
Rate of Reaction

What is the speed of a reaction process?

Why?

Chemical reactions occur at different speeds. Some are ultra-rapid, whereas others require patience. For example, you can burn petrol in just a few seconds, but significant amounts depending on the conditions, it takes months for limestone to change the speed of a reaction, as does dissolving or solidifying it. How the rate of reaction is measured during a reaction.

Model 1 – Concentration versus Time Graph



- The graph in Model 1. How does the concentration of a specific chemical species change over time?
 - What unit is the concentration measured in?
The concentration is measured in mol/l.
 - What unit is the time measured in?
The time is measured in seconds.
- Consider the data in Model 1:
 - What specific concentration of the species when the chemical reaction was followed?
The initial concentration of the chemical species for the reaction was 0.100M.
 - Did the concentration of the species increase or decrease over time?
The concentration of the chemical species decreased over time.
 - Was the reaction a reaction or product for the reaction? Justify your reasoning.
The chemical species was a reactant because it was used up during the reaction. Its concentration went down.

Collision Theory Pogil Answer Key

R Bogdan



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