

Basic Atomic Structure Key

Complete the following using your Periodic Table as reference.

1. 19 Atomic number of potassium
2. 30.97 Mass number of phosphorus
3. 7 Protons in nitrogen
4. 6 Electrons in carbon
5. 12 Neutrons in sodium
6. 13 Atomic number of aluminum
7. 40.08 Mass number of calcium
8. 18 Protons in argon
9. 23 Electrons in vanadium
10. 24 Neutrons in scandium
11. 61 Atomic number of promethium
12. 152.0 Mass number of europium
13. 65 Protons in terbium
14. 47 Electrons in silver
15. 78 Neutrons in cesium
16. 35 Atomic number of bromine
17. 186.2 Mass number of rhenium
18. 77 Protons in iridium
19. 78 Electrons in platinum
20. 118 Neutrons in gold

Isotopes

1. There are 5 naturally occurring isotopes of zinc: . None of these isotopes has the atomic mass of 65.41, as listed on the periodic table for zinc. Explain why.

The mass number listed on the periodic table is an average abundance of all the isotopes.

2. Two isotopes of gallium occur naturally, . Which isotope is more abundant? How do you know?

The mass number of gallium on the periodic table is listed as 69.72, which is closer to 69 than 71, which indicates that it is the more abundant isotope.

Chapter Atomic Structure Wordwise Answers

Mark Steyvers



Chapter Atomic Structure Wordwise Answers:

Whispering the Strategies of Language: An Mental Quest through **Chapter Atomic Structure Wordwise Answers**

In a digitally-driven earth where monitors reign great and immediate connection drowns out the subtleties of language, the profound secrets and psychological nuances hidden within phrases usually go unheard. However, situated within the pages of **Chapter Atomic Structure Wordwise Answers** a captivating fictional value pulsating with raw emotions, lies a fantastic quest waiting to be undertaken. Published by a skilled wordsmith, this marvelous opus attracts readers on an introspective journey, gently unraveling the veiled truths and profound affect resonating within the cloth of each word. Within the emotional depths of the poignant review, we will embark upon a honest exploration of the book is key styles, dissect its interesting writing style, and succumb to the strong resonance it evokes serious within the recesses of readers hearts.

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