

Experimental Conditions			
	Reactants	Products	
Before reaction	 2 molecules N_2	 3 molecules H_2	0 molecules NH_3
After reaction	 1 molecule N_2	0 molecules H_2	 2 molecules NH_3

- In this reaction, only the hydrogen is completely used up.
- H_2 is the **limiting reagent**, or the reactant that determines the amount of product that can be formed by a reaction.

12 3 Limiting Reagent And Percent Yield Answers

Mary Millhollon, Richard H. Langley



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introduction to general organic and biological chemistry Not only is Raymond's text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry

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