

## Chapter Test B

### Chapter Chemical Equations and Reactions

Write the full name (including units) for the term at the end of the sentence. Please show your work. Please write answers without answers with symbols.

- The gas volume of a diatomic molecule is  $2.0 \times 10^3$  L. How many molecules of gas are there?
  - gas
  - grams
  - moles
  - atoms
- The volume of a gas is  $2.0 \times 10^3$  L. How many molecules of gas are there?
  - moles
  - atoms
  - grams
  - grams of gas
- The volume of a gas is  $2.0 \times 10^3$  L. How many molecules of gas are there?
  - atoms
  - grams of gas
  - moles
  - grams
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  - atoms
  - grams of gas
  - grams
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  - atoms
  - grams of gas
  - grams
  - grams of gas

# Chapter Test B Chemistry Chapter Key

**J Rink**



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