

12.1

CHEMICAL REACTIONS

Section Review

Objectives

1. Identify substances using a chemical change.
2. Identify and predict chemical reactions using the laws of conservation.
3. Explain the law of conservation of mass in chemical reactions.

Vocabulary

- | | | |
|-----------------------------|--------------------|---------------------------------------|
| 1. chemical change | 2. reactant | 3. product |
| 4. chemical reaction | 5. product | 6. law of conservation of mass |

Part A Completion

Write the complete formula in each blank space. Use the chemical symbols and the atomic symbols given in parentheses. Balance the chemical equations with the proper coefficients in parentheses.

- | | |
|--------------------------------------------------------------------------------------|--------------------------------------------------------------|
| 1. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ | 6. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ |
| 2. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ | 7. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ |
| 3. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ | 8. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ |
| 4. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ | 9. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ |
| 5. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ | 10. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ |
| 11. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ | 12. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ |
| 13. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ | 14. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ |
| 15. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ | 16. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ |
| 17. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ | 18. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ |
| 19. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ | 20. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ |

Part B True-False

Write true or false in the blank space. If false, write the correct formula in the space provided.

1. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
2. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
3. $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

2neutralization Reactions Section Review

G Psacharopoulos



2neutralization Reactions Section Review:

Unveiling the Magic of Words: A Overview of "**2neutralization Reactions Section Review**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**2neutralization Reactions Section Review**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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