

**CHEMISTRY**  
**PAPER – 2**  
**(PRACTICAL)**

*(Maximum Marks: 30)*

*(Time allowed: Three hours)*

*(Candidates are allowed additional 15 minutes for only reading the paper.  
They must NOT start writing during this time.)*

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**ALL ANSWERS MUST BE WRITTEN IN THE ANSWER BOOKLET PROVIDED SEPARATELY.**

*Question 1 is an oxidation–reduction titration in which sufficient working details are given.  
All essential working must be shown.*

*Question 2 is an exercise dealing with identification of organic compounds.  
Credit will be given for precise observations recorded and for well–drawn deductions.*

*Question 3 is an exercise in qualitative analysis.*

---

*Read the questions carefully and follow the given instructions.*

*Attempt all questions.*

*All working, including rough work, should be done on the same sheet as the rest of the answer.*

*The intended marks for questions or parts of questions are given in brackets [ ].*

*Mathematical Tables are provided.*

---

*Attempt all questions.*

**Question 1**

[7]

You are provided with two solutions as follows:

- C-10 is a solution containing 1.95 gms of potassium manganate (VII),  $\text{KMnO}_4$  per litre.
  - C-11 is a solution prepared by dissolving 23.2 gms of hydrated ammonium iron (II) sulphate crystals,  $(\text{NH}_4)_2\text{SO}_4 \cdot \text{FeSO}_4 \cdot x\text{H}_2\text{O}$  per litre.
- 

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**Turn over**

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# 2physical Science Practical Papers

**Ying Liu**



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