

General Instructions :

Read the following instructions very carefully and strictly follow them :

- (i) This question paper comprises **15** questions. **All** questions are compulsory.
- (ii) This question paper is divided into **three** sections – **A, B** and **C**.
- (iii) **Section A** – Questions No. **1** to **7** are short answer type questions. Each question carries **2** marks.
- (iv) **Section B** – Questions No. **8** to **13** are also short answer type questions. Each question carries **3** marks.
- (v) **Section C** – Questions No. **14** and **15** are case-based questions. Each question carries **4** marks.
- (vi) Internal choices have been provided in some questions. Only one of the alternatives has to be attempted.

SECTION A

1. (a) Write the molecular formula of the following carbon compounds :
 - (i) Methane
 - (ii) Propane
- (b) Carbon compounds have low melting and boiling points. Why ? 2
2. The electrons in the atoms of two elements X and Y are distributed in three shells having 1 and 7 electrons respectively in their outermost shells.
 - (a) Write the group numbers of these elements in the Modern Periodic Table.
 - (b) Write the molecular formula of the compound formed when X and Y combine with each other.
 - (c) Which of the two is electropositive ? 2
3. (a) Which of the following flowers will have higher possibility of self-pollination ?

Mustard, Papaya, Watermelon, Hibiscus
- (b) List the two reproductive parts of a bisexual flower. 2
4. Which one of the two multicellular organisms — Spirogyra and Planaria reproduces by regeneration and why ? Give an example of any other organism which can also reproduce by the same process. 2



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Table of Contents 2physical Science Grade Question Paper

1. Understanding the eBook 2physical Science Grade Question Paper
 - The Rise of Digital Reading 2physical Science Grade Question Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying 2physical Science Grade Question Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 2physical Science Grade Question Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from 2physical Science Grade Question Paper
 - Personalized Recommendations
 - 2physical Science Grade Question Paper User Reviews and Ratings
 - 2physical Science Grade Question Paper and Bestseller Lists
5. Accessing 2physical Science Grade Question Paper Free and Paid eBooks
 - 2physical Science Grade Question Paper Public Domain eBooks
 - 2physical Science Grade Question Paper eBook Subscription Services
 - 2physical Science Grade Question Paper Budget-Friendly Options
6. Navigating 2physical Science Grade Question Paper eBook Formats

- ePub, PDF, MOBI, and More
 - 2physical Science Grade Question Paper Compatibility with Devices
 - 2physical Science Grade Question Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 2physical Science Grade Question Paper
 - Highlighting and Note-Taking 2physical Science Grade Question Paper
 - Interactive Elements 2physical Science Grade Question Paper
 8. Staying Engaged with 2physical Science Grade Question Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 2physical Science Grade Question Paper
 9. Balancing eBooks and Physical Books 2physical Science Grade Question Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 2physical Science Grade Question Paper
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine 2physical Science Grade Question Paper
 - Setting Reading Goals 2physical Science Grade Question Paper
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of 2physical Science Grade Question Paper
 - Fact-Checking eBook Content of 2physical Science Grade Question Paper
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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