

Biochemistry Study Guide

- Monomers: stable molecule but readily bonds with other monomers
- Polymers: monomers in a chain formation of that group
- Dehydration synthesis: the reaction to build polymers from monomers (polymerization) (water molecule is released: H and OH that bonds)
 - Anabolic: build complex molecules from simpler ones => Endergonic (construction) => Consumption of Energy => $+\Delta G$ => Nonspontaneous
- Hydrolysis: the reaction to break down polymers into monomers (water molecule is a reactant)
 - Catabolic: break down complex molecules to simpler ones => Exergonic (releasing energy from the energy stored in the bonds) => Release of Energy => $-\Delta G$ => Spontaneous
- Directionality influences the Structure and Function of the Polymer
 - Amino Acids: N-terminus and C-terminus determine from which side can more amino acids be added and thus the structure and function of the protein is influenced
 - Nucleic Acids: linear sequence of nucleotides that have ends, defined by the 3' hydroxyl and 5' phosphates of the sugar in the nucleotide. During DNA and RNA synthesis, nucleotides are added to the 3' end of the growing strand, resulting in the formation of a covalent bond between nucleotides.
- Uniqueness of Carbon:
 - Atom is small in size with the valence electrons close to the nucleus, so it creates stable bonds
 - 4 valence electrons allows it to bond up to 4 different elements
 - Multiple types of Carbon Bonds: C-C, C=C, $C\equiv C$, $C\equiv C$

Macromolecule Class	Elemental composition	Monomer (or subcomponents)	Polymer	Shape / structure	Biological function/s	Key example/s	Structure-function relationship example/s	Other notes or properties
Carbohydrates	CHO	monosaccharide	polysaccharide	(carbon) rings	Energy storage (long and short), structure	Cellulose, starch (lots of glucose), glycogen	Cellulose: Alpha Glucose chain is a rigid structure	Glycosidic bond Covalent bonds C-C Most end in -ose (classified by # of carbons)
Lipids	CHO	Subcomponents: glycerol head and fatty acids	N/A	Starfish (fatty acids attached to glycerol head)	Long term energy storage, insulation/cushioning, membranes	Saturated fats, unsaturated fats, phospholipids	Phospholipids (amphipathic) = spherical membranes	Ester Linkage bond Nonpolar

Biochemistry Study Guides

Padhraic Smyth



Biochemistry Study Guides:

Embracing the Melody of Phrase: An Psychological Symphony within **Biochemistry Study Guides**

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