

EN		ENTREPRENEURSHIP	TIME - 90 MIN. MAX. MARKS - 100
ROLL NUMBER	LOCAL SELECTION	SELECTION NUMBER	MARKS
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CANDIDATE'S SIGNATURE		CANDIDATE'S FINGERPRINT	

1. Read the following passage carefully and answer the questions that follow. Write your answers in the spaces provided.

2. The passage is about the importance of entrepreneurship in the Indian economy. It discusses the role of entrepreneurs in creating jobs and driving economic growth.

3. The passage also highlights the challenges faced by entrepreneurs, such as lack of capital and government support.

4. The passage concludes by stating that entrepreneurship is essential for the development of a nation.

5. The passage is written in a formal and objective tone.

6. The passage is divided into several paragraphs, each focusing on a different aspect of entrepreneurship.

7. The passage uses simple and clear language to convey its message.

8. The passage is well-organized and easy to read.

9. The passage is a good example of expository writing.

10. The passage is a good example of how to write a formal letter.

Answer any 10 questions out of 12.	Mark each correct answer as follows:
<input type="radio"/> (a) <input type="radio"/> (b) <input type="radio"/> (c) <input type="radio"/> (d)	<input checked="" type="radio"/> (a) <input type="radio"/> (b) <input type="radio"/> (c) <input type="radio"/> (d)

11. The passage is a good example of how to write a formal letter.

12. The passage is a good example of how to write a formal letter.

13. The passage is a good example of how to write a formal letter.

14. The passage is a good example of how to write a formal letter.

15. The passage is a good example of how to write a formal letter.



PART - A
 APPLIED SCIENCE

1. Identify the odd one out.

(a) Brass	(b) Steel
(c) Bronze	(d) Copper
2. The pH of water is 7. The number of hydrogen ions in 100 ml of water present in it is _____.

[Question Paper](#)

[Key Answer](#)

Diploma Cet Question Papers

L Cohen



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web atoms of the same element i e same atomic number Z that have different numbers of neutrons are called isotopes for example ${}^{12}_6\text{C}$ of the carbon atoms on earth have 6 neutrons and 6 protons in their nuclei about 1% of the carbon atoms have 7

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