

Chemistry

Name: _____ Date: _____

Write the following chemical reactions and balance them.

1. $\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$

2. $\text{H}_2 + \text{Cl}_2 \rightarrow \text{HCl}$

3. $\text{Ca} + \text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2 + \text{H}_2$

4. $\text{Mg} + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$

5. $\text{Al} + \text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + \text{H}_2$

6. $\text{Cu} + \text{AgNO}_3 \rightarrow \text{Cu(NO}_3)_2 + \text{Ag}$

7. $\text{Fe} + \text{CuSO}_4 \rightarrow \text{FeSO}_4 + \text{Cu}$

8. $\text{Zn} + \text{H}_2\text{SO}_4 \rightarrow \text{ZnSO}_4 + \text{H}_2$

9. $\text{Na} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{H}_2$

10. $\text{K} + \text{H}_2\text{O} \rightarrow \text{KOH} + \text{H}_2$

11. $\text{Ca} + \text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2 + \text{H}_2$

12. $\text{Mg} + \text{H}_2\text{O} \rightarrow \text{Mg(OH)}_2 + \text{H}_2$

13. $\text{Al} + \text{H}_2\text{O} \rightarrow \text{Al(OH)}_3 + \text{H}_2$

14. $\text{Fe} + \text{H}_2\text{O} \rightarrow \text{Fe}_3\text{O}_4 + \text{H}_2$

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Robert R. Redfield



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