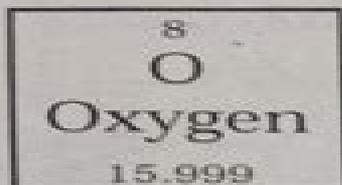


The Atoms Family  
Atomic Math Challenge

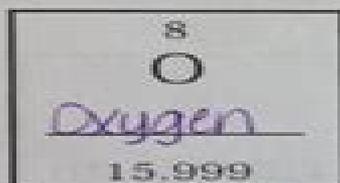
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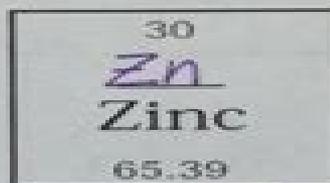
← Atomic Number  
← Symbol  
← Name  
← Atomic Mass

Atomic number equals  
the number of  
protons or electrons

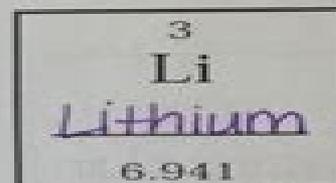
Atomic mass equals  
the number of  
protons + neutrons



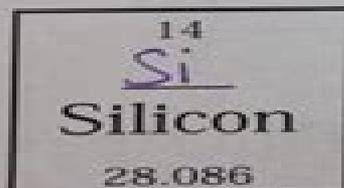
Atomic # = 8  
Atomic Mass = 16  
# of Protons = 8  
# of Neutrons = 8  
# of Electrons = 8



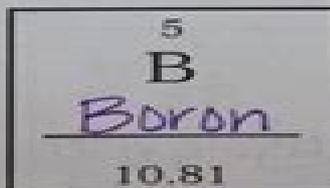
Atomic # = 30  
Atomic Mass = 65  
# of Protons = 30  
# of Neutrons = 35  
# of Electrons = 30



Atomic # = 3  
Atomic Mass = 7  
# of Protons = 3  
# of Neutrons = 4  
# of Electrons = 3



Atomic # = 14  
Atomic Mass = 28  
# of Protons = 14  
# of Neutrons = 14  
# of Electrons = 14



Atomic # = 5  
Atomic Mass = 11  
# of Protons = 5  
# of Neutrons = 6  
# of Electrons = 5



Atomic # = 35  
Atomic Mass = 80  
# of Protons = 35  
# of Neutrons = 45  
# of Electrons = 35

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**Ensheng Dong**



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