

1.  $\underline{\quad}$  Zn +  $\underline{2}$  HCl  $\rightarrow$   $\underline{\quad}$  ZnCl<sub>2</sub> +  $\underline{\quad}$  H<sub>2</sub>
2.  $\underline{4}$  Fe +  $\underline{13}$  O<sub>2</sub>  $\rightarrow$   $\underline{2}$  Fe<sub>2</sub>O<sub>3</sub>
3.  $\underline{4}$  P +  $\underline{5}$  O<sub>2</sub>  $\rightarrow$   $\underline{\quad}$  P<sub>4</sub>O<sub>10</sub>
4.  $\underline{\quad}$  N<sub>2</sub> +  $\underline{3}$  H<sub>2</sub>  $\rightarrow$   $\underline{2}$  NH<sub>3</sub>
5.  $\underline{\quad}$  PbO<sub>2</sub>  $\rightarrow$   $\underline{\quad}$  Pb +  $\underline{\quad}$  O<sub>2</sub>
6.  $\underline{2}$  C<sub>6</sub>H<sub>6</sub> +  $\underline{15}$  O<sub>2</sub>  $\rightarrow$   $\underline{12}$  CO<sub>2</sub> +  $\underline{6}$  H<sub>2</sub>O
7.  $\underline{\quad}$  C<sub>2</sub>H<sub>5</sub>OH +  $\underline{3}$  O<sub>2</sub>  $\rightarrow$   $\underline{2}$  CO<sub>2</sub> +  $\underline{3}$  H<sub>2</sub>O
8.  $\underline{3}$  Fe +  $\underline{4}$  H<sub>2</sub>O  $\rightarrow$   $\underline{\quad}$  Fe<sub>3</sub>O<sub>4</sub> +  $\underline{4}$  H<sub>2</sub>
9.  $\underline{\quad}$  SiO<sub>2</sub> +  $\underline{4}$  HF  $\rightarrow$   $\underline{\quad}$  SiF<sub>4</sub> +  $\underline{2}$  H<sub>2</sub>O
10.  $\underline{3}$  CO +  $\underline{\quad}$  FeO<sub>4</sub>  $\rightarrow$   $\underline{\quad}$  FeO +  $\underline{3}$  CO<sub>2</sub>
11.  $\underline{\quad}$  NH<sub>4</sub>NO<sub>2</sub>  $\rightarrow$   $\underline{\quad}$  N<sub>2</sub> +  $\underline{2}$  H<sub>2</sub>O
12.  $\underline{\quad}$  N<sub>2</sub>H<sub>4</sub> +  $\underline{\quad}$  O<sub>2</sub>  $\rightarrow$   $\underline{\quad}$  N<sub>2</sub> +  $\underline{2}$  H<sub>2</sub>O
13.  $\underline{3}$  NO<sub>2</sub> +  $\underline{\quad}$  H<sub>2</sub>O  $\rightarrow$   $\underline{2}$  HNO<sub>3</sub> +  $\underline{\quad}$  NO
14.  $\underline{2}$  KClO<sub>3</sub>  $\rightarrow$   $\underline{2}$  KCl +  $\underline{3}$  O<sub>2</sub>
15.  $\underline{\quad}$  C<sub>12</sub>H<sub>22</sub>O<sub>11</sub> +  $\underline{12}$  O<sub>2</sub>  $\rightarrow$   $\underline{12}$  CO<sub>2</sub> +  $\underline{11}$  H<sub>2</sub>O
16.  $\underline{4}$  NH<sub>3</sub> +  $\underline{3}$  O<sub>2</sub>  $\rightarrow$   $\underline{6}$  H<sub>2</sub>O +  $\underline{2}$  N<sub>2</sub>
17.  $\underline{\quad}$  CH<sub>3</sub>COOH +  $\underline{\quad}$  O<sub>2</sub>  $\rightarrow$   $\underline{2}$  CO<sub>2</sub> +  $\underline{2}$  H<sub>2</sub>O
18.  $\underline{2}$  C<sub>2</sub>H<sub>6</sub> +  $\underline{7}$  O<sub>2</sub>  $\rightarrow$   $\underline{4}$  CO<sub>2</sub> +  $\underline{6}$  H<sub>2</sub>O
19.  $\underline{\quad}$  C<sub>6</sub>H<sub>6</sub> +  $\underline{10}$  O<sub>2</sub>  $\rightarrow$   $\underline{8}$  CO<sub>2</sub> +  $\underline{4}$  H<sub>2</sub>O
20.  $\underline{2}$  C<sub>4</sub>H<sub>10</sub> +  $\underline{13}$  O<sub>2</sub>  $\rightarrow$   $\underline{8}$  CO<sub>2</sub> +  $\underline{10}$  H<sub>2</sub>O
21.  $\underline{\quad}$  N<sub>2</sub> +  $\underline{3}$  H<sub>2</sub>  $\rightarrow$   $\underline{2}$  NH<sub>3</sub>
22.  $\underline{3}$  Fe +  $\underline{4}$  H<sub>2</sub>O  $\rightarrow$   $\underline{\quad}$  Fe<sub>3</sub>O<sub>4</sub> +  $\underline{4}$  H<sub>2</sub>
23.  $\underline{3}$  Fe +  $\underline{2}$  O<sub>2</sub>  $\rightarrow$   $\underline{\quad}$  Fe<sub>3</sub>O<sub>4</sub>
24.  $\underline{3}$  NH<sub>4</sub>OH +  $\underline{\quad}$  AlCl<sub>3</sub>  $\rightarrow$   $\underline{\quad}$  Al(OH)<sub>3</sub> +  $\underline{3}$  NH<sub>4</sub>Cl
25.  $\underline{\quad}$  Na<sub>2</sub>CO<sub>3</sub> +  $\underline{\quad}$  H<sub>2</sub>O +  $\underline{\quad}$  CO<sub>2</sub>  $\rightarrow$   $\underline{2}$  NaHCO<sub>3</sub>
26.  $\underline{\quad}$  MnO<sub>2</sub> +  $\underline{4}$  HCl  $\rightarrow$   $\underline{\quad}$  MnCl<sub>2</sub> +  $\underline{2}$  H<sub>2</sub>O +  $\underline{\quad}$  Cl<sub>2</sub>
27.  $\underline{\quad}$  PBr<sub>3</sub> +  $\underline{3}$  H<sub>2</sub>O  $\rightarrow$   $\underline{\quad}$  H<sub>3</sub>PO<sub>3</sub> +  $\underline{3}$  HBr
28.  $\underline{2}$  ZnS +  $\underline{3}$  O<sub>2</sub>  $\rightarrow$   $\underline{2}$  ZnO +  $\underline{\quad}$  SO<sub>2</sub>
29.  $\underline{2}$  NH<sub>3</sub> +  $\underline{3}$  O<sub>2</sub>  $\rightarrow$   $\underline{\quad}$  N<sub>2</sub> +  $\underline{6}$  H<sub>2</sub>O
30.  $\underline{\quad}$  Na<sub>2</sub>B<sub>4</sub>O<sub>7</sub> +  $\underline{\quad}$  H<sub>2</sub>SO<sub>4</sub> +  $\underline{5}$  H<sub>2</sub>O  $\rightarrow$   $\underline{\quad}$  Na<sub>2</sub>SO<sub>4</sub> +  $\underline{4}$  H<sub>3</sub>BO<sub>3</sub>
31.  $\underline{2}$  NH<sub>3</sub> +  $\underline{2}$  O<sub>2</sub>  $\rightarrow$   $\underline{\quad}$  NO +  $\underline{3}$  H<sub>2</sub>O
32.  $\underline{4}$  HNO<sub>3</sub>  $\rightarrow$   $\underline{2}$  H<sub>2</sub>O +  $\underline{4}$  NO<sub>2</sub> +  $\underline{\quad}$  O<sub>2</sub>
33.  $\underline{4}$  HCl +  $\underline{2}$  HNO<sub>3</sub>  $\rightarrow$   $\underline{2}$  Cl<sub>2</sub> +  $\underline{2}$  NOCl +  $\underline{3}$  H<sub>2</sub>O
34.  $\underline{2}$  Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> +  $\underline{6}$  SiO<sub>2</sub>  $\rightarrow$   $\underline{6}$  CaSiO<sub>3</sub> +  $\underline{\quad}$  P<sub>4</sub>O<sub>10</sub>
35.  $\underline{\quad}$  C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>  $\rightarrow$   $\underline{2}$  C<sub>6</sub>H<sub>5</sub>OH +  $\underline{2}$  CO<sub>2</sub>
36.  $\underline{2}$  C<sub>8</sub>H<sub>18</sub> +  $\underline{25}$  O<sub>2</sub>  $\rightarrow$   $\underline{16}$  CO<sub>2</sub> +  $\underline{18}$  H<sub>2</sub>O
37.  $\underline{2}$  CH<sub>4</sub> +  $\underline{3}$  O<sub>2</sub>  $\rightarrow$   $\underline{2}$  CO +  $\underline{4}$  H<sub>2</sub>O
38.  $\underline{2}$  C<sub>2</sub>H<sub>2</sub> +  $\underline{3}$  O<sub>2</sub>  $\rightarrow$   $\underline{4}$  CO<sub>2</sub> +  $\underline{2}$  H<sub>2</sub>O
39.  $\underline{6}$  H<sub>2</sub>O +  $\underline{6}$  CO<sub>2</sub>  $\rightarrow$   $\underline{\quad}$  C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> +  $\underline{6}$  O<sub>2</sub>
40.  $\underline{\quad}$  Cu +  $\underline{4}$  HNO<sub>3</sub>  $\rightarrow$   $\underline{\quad}$  Cu(NO<sub>3</sub>)<sub>2</sub> +  $\underline{2}$  NO +  $\underline{2}$  H<sub>2</sub>O

# Balancing Equations Ch 7 Worksheet Answers Chapter 7

**Chris McMullen**



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