

(c) This reaction takes place in a number of steps, some of which are shown below.



(i) State the type of reaction occurring in step 1 and the conditions needed for this step.

(2)

Type _____

Conditions _____

(ii) Complete the equation below for the third step of the reaction, and show the movement of electrons using the appropriate arrows.

(3)



(iii) Write equations for **two** termination steps in this reaction.

(2)

(d) Ethane can be cracked in industry. Write an equation for the cracking of ethane.

(1)

(e) Suggest **two** reasons why cracking of larger alkane molecules is important in industry.

(2)

Reason 1: _____

Reason 2: _____

(Total for Question 20 = 16 marks)

TOTAL FOR SECTION B = 60 MARKS
TOTAL FOR PAPER = 80 MARKS



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Michael Seilmaier



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