Empirical and Molecular Formula Worksheet ANSWER KEY

Write the empirical formula for the following compounds.

1) C_2H_2 **CH**

6) C_6H_{14} **C_3H_7**

7) WO_2 **WO_2**

8) $C_2H_6O_2$ **CH_3O**

9) $X_{39}Y_{13}$ **X_3Y**

6)   A compound with an empirical formula of C_2OH_4 and a molar mass of 88 grams per mole. What is the molecular formula of this compound?   **$C_4O_2H_8$**

7)   A compound with an empirical formula of C_4H_4O and a molar mass of 136 grams per mole. What is the molecular formula of this compound?   **$C_8H_8O_2$**

8)   A compound with an empirical formula of CFBrO and a molar mass of 253.8 grams per mole. What is the molecular formula of this compound?   **$C_2F_2Br_2O_2$**

9)   A compound with an empirical formula of C_2H_8N and a molar mass of 46 grams per mole. What is the molecular formula of this compound?   **C_2H_8N**

10)   A well-known reagent in analytical chemistry, dimethylglyoxime, has the empirical formula C_2H_4NO. If its molar mass is 116.1 g/mol, what is the molecular formula of the compound?   **$C_4H_8N_2O_2$**

12.   A certain blue solid contains 36.84% N and 63.16% O. What is the empirical formula of this compound? **The ratios are $N_{1.00}O_{1.50}$. Since 1.50 is not close to a whole number, we multiply *both* subscripts by 2. The empirical formula is thus N_2O_3. (The name is dinitrogen trioxide.)**

13.   A sample of indium chloride weighing 0.5000 g is found to contain 0.2404 g of chlorine. What is the empirical formula of the indium compound? **$InCl_3$**

14.   An unknown compound was found to have a percent composition as follows: 47.0 % potassium, 14.5 % carbon, and 38.5 % oxygen. What is its empirical formula? If the true molar mass of the compound is 166.22 g/mol, what is its molecular formula? **$K_2C_2O_4$**

15.   Rubbing alcohol was found to contain 60.0 % carbon, 13.4 % hydrogen, and the remaining mass was due to oxygen. What is the empirical formula of rubbing alcohol? **C_3H_8O**

Calculating Empirical Formulas 6 3 Practice Answer Key

Raffaela Di Napoli

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