

PRACTICE EXAM 2  
 ITEM 1411 (Chap. 1, 2, 3, 4, 5)

1. Which of the following is a physical change?  
 (a) rusting a nail (b) fruits ripen (c) baking a cake  
 (d) ice melts (e) milk turns sour
2. Which of the following is a homogeneous mixture?  
 (a) water and sand (b) sand, alcohol, and sugar  
 (c) oil and vinegar (d) milk of magnesium (e) air
3. Which of the following is a correct?  
 (a) sodium, S (b) mercury, Mg (c) tin, N, Sn  
 (d) potassium, P (e) gold, Au
4. What is the equivalent temperature for 105 Fahrenheit in Kelvin?  
 (a) 41 (b) 314 (c) 378 (d) 457 (e) -32
5. A sample of carbon tetrachloride has a mass of 39.75 g and a volume of 25.0 mL. What is its density (g/mL)?  
 (a) 1.59 (b) 1.74 (c) 2.36 (d) 2.89 (e) 3.50  
 $d = \frac{m}{V} = \frac{39.75}{25.0 \text{ mL}}$
6. How many significant figures are there in the measurement of  $1.0356 \times 10^4$ ?  
 (a) 1 (b) 2 (c) 3 (d) 4 (e) 5
7. How many significant figures are appropriate to show in the result after carrying out the operation below?  
 $(13.7 + 0.027) + 8.221 = ?$   $\times, \div$  least s.f.  $+, -$  least decimal
- (a) 1 (b) 2 (c) 3 (d) 4 (e) 5
8. How many protons, electrons, and neutrons are there in  $^{130}\text{Ba}^{2+}$ ?  
 (a) 56p, 56e, 130n (b) 56p, 54e, 130n (c) 56p, 58e, 74n  
 (d) 56p, 54e, 74n (e) 54p, 56e, 74n
9. Which of the following pairs would have similar chemical and physical properties?  
 (a) Na, Mg (b) H, Li (c) He, Ne (d) Cr, Ca (e) Cl, O
10. Which of the following is the molecular formula?  
 (a) NaCl (b)  $\text{C}_2\text{F}_4$  (c) HCl (d)  $\text{AlBr}_3$  (e) NaCN
11. Which of the following is the empirical formula?  
 (a)  $\text{C}_2\text{H}_4$  (b)  $\text{CH}_2\text{COOH}$  (c)  $\text{P}_2\text{O}_{10}$  (d)  $\text{NH}_4\text{NO}_3$  (e)  $\text{Ag}_2\text{CO}_3$
12. Which of the following is a correct match?  
 (a)  $\text{BaCl}_2$ , barium dichloride (b)  $\text{CS}_2$ , carbon sulfide  
 (c)  $\text{CuF}_2$ , copper (I) fluoride (d)  $\text{HNO}_2$ , nitrous acid  
 (e)  $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ , magnesium sulfate heptahydrate
13. Which of the following pairs is a correct match?  
 (a)  $\text{Ca}_3(\text{PO}_4)_2$ , calcium (II) phosphate (c)  $\text{Mg}_2\text{O}_3$ , magnesium oxide  
 (b)  $(\text{NH}_4)_2\text{SO}_4$ , ammonium sulfate (d)  $\text{H}_2\text{CO}_3$ , carbon acid  
 (e)  $\text{Cu}(\text{NO}_3)_2$ , copper dinitrate

# Chemistry1412 Lab Manual Answers

**RS Peters**



**Chemistry1412 Lab Manual Answers:**

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Chemistry1412 Lab Manual Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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