

# States of Matter and Phase Changes

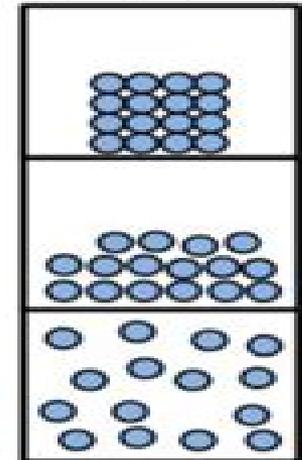
Matter can exist in three major states:

**→ Solid** → Low temperature and low energy.  
Has a definite volume and a definite shape.

**→ Liquid** → Takes the shape of its container.  
Has a definite volume, but no definite composition.

**→ Gas** → High amount of energy and disorder.  
Has no definite shape and no definite composition.

Draw each state below:



**Instructions:** Identify the **phase changes** below and whether energy is **increasing** or **decreasing**.

1)		<p>Energy</p> <p>→</p> <p>Increase <b>or</b> decrease</p>		Phase Change: <u>Freezing</u>
2)		<p>Energy</p> <p>→</p> <p>Increase <b>or</b> decrease</p>		Phase Change: <u>Condensation</u>
3)		<p>Energy</p> <p>→</p> <p>Increase <b>or</b> decrease</p>		Phase Change: <u>Melting</u>

# Chemistry Study Guide Phase Change Answer Questions

**Sabine Zange**



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