

Intro. to Equilibrium

Introduction: Forward and Reverse Processes

Reversible processes are chemical reactions that proceed in both directions. For example, the decomposition of hydrogen peroxide into hydrogen gas and water is a reversible process:



If you start with only the reactants, the reaction will proceed in the forward direction as the system begins to reach equilibrium. If you start with only the products, the reaction will proceed in the reverse direction as the system begins to reach equilibrium. At equilibrium, the rates of the forward and reverse processes are equal.

Reversible processes are also called dynamic equilibrium processes. At equilibrium, the system is still changing, but the overall composition remains constant. This is because the forward and reverse processes are occurring at equal rates.



The following equation represents the reaction above at equilibrium:



The reaction is reversible, so it can proceed in both directions. At equilibrium, the rates of the forward and reverse processes are equal. This is because the forward and reverse processes are occurring at equal rates.

Critical Thinking Questions

- Consider the reversible reaction between hydrogen and oxygen. In the beginning of the reaction, there are only hydrogen and oxygen molecules. As the reaction proceeds, the number of hydrogen and oxygen molecules decreases, and the number of water molecules increases. When equilibrium is reached, the number of hydrogen and oxygen molecules is constant, and the number of water molecules is constant.
 - What is the effect of the change in the number of water molecules on the reaction? Is the reaction reversible? If so, in which direction? Why? (Hint: Think about the number of water molecules that are formed by the reaction.)
 - What is the effect of the change in the number of hydrogen and oxygen molecules on the reaction? Is the reaction reversible? If so, in which direction? Why? (Hint: Think about the number of hydrogen and oxygen molecules that are formed by the reaction.)

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Chemquest 45 Intro To Equilibrium Answer Key

AW Rasmussen



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