

Name: \_\_\_\_\_

## Research Paper Rubric

|                          | 4   | 3  | 2  | 1  |
|--------------------------|---|--|--|--|
| <b>Content and Focus</b> | <ul style="list-style-type: none"> <li>• Exceptionally clear, focused, interesting thesis.</li> <li>• Strong, rich supporting details and examples that prove thesis.</li> <li>• A meaningful conclusion explaining the importance of the research and how it can be used.</li> </ul>   | <ul style="list-style-type: none"> <li>• Clear thesis which maintains a consistent focus from beginning to end.</li> <li>• Specific supporting details are present.</li> <li>• A clear conclusion as to why the research is important.</li> </ul>  | <ul style="list-style-type: none"> <li>• Contains thesis but with inconsistent focus.</li> <li>• Generalized supporting details that prove thesis.</li> <li>• Conclusion tends to summarize research.</li> </ul>   | <ul style="list-style-type: none"> <li>• Thesis statement lacks clarity and focus.</li> <li>• Inadequate or missing supporting details.</li> <li>• Missing a summarizing conclusion.</li> </ul>                                      |
| <b>Organization</b>      | <ul style="list-style-type: none"> <li>• Strong introduction and conclusion.</li> <li>• Consistent and coherent logical progression.</li> <li>• Uses clear and skillful transitions.</li> </ul>   | <ul style="list-style-type: none"> <li>• Clear introduction and conclusion.</li> <li>• Illustrates some consistency and shows some logical progression.</li> <li>• Uses clear transitions.</li> </ul>  | <ul style="list-style-type: none"> <li>• Introduction and conclusion is present but not clear.</li> <li>• Show some attempt of consistency and order. Paper shows attempt of transitions between paragraphs.</li> </ul>  | <ul style="list-style-type: none"> <li>• Unable to clearly identify introduction and conclusion.</li> <li>• Lack of consistency and order.</li> <li>• Paper shows little or no attempt of transitions between paragraphs.</li> </ul> |
| <b>Style</b>             | <ul style="list-style-type: none"> <li>• Written in formal language (avoids slang, completely).</li> <li>• Elaborate and colorful language.</li> <li>• Consistently strong and varied sentence structure.</li> <li>• Direct quotes support student's ideas.</li> <li>• Paper written in student's own words.</li> </ul>                                   | <ul style="list-style-type: none"> <li>• Majority of paper written in formal language.</li> <li>• Language appropriate to topic.</li> <li>• Words convey intended message.</li> <li>• Direct quotes support student's ideas.</li> <li>• Majority of paper written in student's own words.</li> </ul>   | <ul style="list-style-type: none"> <li>• Some use of formal language recognized; informal language is dominant.</li> <li>• Most language is appropriate to topic. Able to get vague idea of message.</li> <li>• Some parts of paper written in student's own words.</li> </ul> | <ul style="list-style-type: none"> <li>• Paper frequently uses informal language.</li> <li>• Language is not appropriate to topic.</li> <li>• Message is unclear.</li> <li>• Majority of paper is plagiarized.</li> </ul>            |
| <b>Sources/Format</b>    | <p>Follows MLA guidelines:</p> <ul style="list-style-type: none"> <li>• Uses 5 or more cited sources.</li> <li>• Sources meet the guidelines for types of sources.</li> <li>• All parenthetical documentation is MLA correct (author's name, pg. #).</li> <li>• Works Cited page is MLA correct.</li> <li>• All researched info is documented.</li> </ul> | <p>Follows MLA Guidelines with few exceptions:</p> <ul style="list-style-type: none"> <li>• 3-4 cited sources used.</li> <li>• Sources meet the guidelines for types of sources.</li> <li>• Few errors noted in parenthetical documentation.</li> <li>• Majority of Works Cited page is MLA correct.</li> <li>• Most research info is Documented.</li> </ul> | <ul style="list-style-type: none"> <li>• Inconsistent use of MLA style guidelines. Less than 3 cited sources used.</li> <li>• Majority of parenthetical documentation done incorrectly.</li> <li>• Random MLA documentation.</li> <li>• Rarely documents sources.</li> </ul>   | <ul style="list-style-type: none"> <li>• Fails to follow MLA style Guidelines.</li> <li>• Less than 3 cited sources used little or no parenthetical documentation.</li> <li>• Works Cited page is not understandable.</li> </ul>     |
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# Science Research Paper Rubric

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more positive the more the polar bear attracts the shared pair of electrons the more positive the partial charge on the penguin

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web answer the following questions consider the following illustration draw the lewis dot structure for hcl if the penguin represents a hydrogen atom and the polar bear represents a chlorine atom what does the ice cream represent in the drawing what do you think the picture is trying to illustrate would hcl be attracted to the charged wand

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web to answer this question you will examine 1 electronegativity 2 nonpolar vs polar molecules 3 electronegativity and bonding electronegativity the h atom and the cl atom in hydrogen chloride

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web 1 how does the comic strip define a polar molecule 2 define electronegativity as you understand it after reading the first two pages of the comic strip 3 what is the artist trying to represent by two polar bears arm wrestling or two penguins arm wrestling 4 what three types of bonds are represented on the third page of the comic strip

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web explain how the iceberg penguins and polar bears represent trends in electronegativity on the periodic table and more study with quizlet and memorize flashcards containing terms like how does the comic define a polar molecule define electronegativity as you understand it after reading the first two pages of the comic interpret the pic

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web so the final representation of the nh<sub>3</sub> molecule using polar bears and penguins is as follows begin array c text polar bear nitrogen text with two lone pairs of electrons text and three penguins hydrogen end array this is a fun and creative way to represent the nh<sub>3</sub> molecule

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web question using polar bears and penguins create an illustration showing an ammonia molecule nh<sub>3</sub> hint you may wish to start with a lewis dot structure

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web polar bears and penguins purpose in this lesson you will be exploring polarity and bonding between atoms in greater detail a comic book will provide new information about these topics and will introduce you to the concept of electronegativity which helps us understand partial charges