

### Chemistry Chapter 6 Review **KEY**

1. What is the difference between a covalent bond and an ionic bond?  
**Covalent** – share electrons to reach a full octet (Example: H—O—H)  
**Ionic** – steals electrons to reach a full octet (Example: Na<sup>+</sup> Cl<sup>-</sup>)
  
2. Determine whether the bond formed between each pair of atoms will be ionic or covalent:
  - a. K and I            **ionic**
  - b. Mg and O        **ionic**
  - c. C and Cl         **covalent**
  - d. Ca and N        **ionic**
  - e. Li and O         **ionic**
  - f. C and H          **covalent**
  
3. Then write the formula of the ionic compound. (Remember that the positive and negative charges must cancel out.)
  - a. Sodium and oxygen                    Na<sup>+</sup>    O<sup>2-</sup>    **Na<sub>2</sub>O**
  - b. Magnesium and chlorine                Mg<sup>2+</sup>   Cl<sup>-</sup>    **MgCl<sub>2</sub>**
  - c. Potassium and fluorine                 K<sup>+</sup>     F<sup>-</sup>     **KF**
  - d. Calcium and carbonate (CO<sub>3</sub><sup>2-</sup>)       Ca<sup>2+</sup>   CO<sub>3</sub><sup>2-</sup>   **CaCO<sub>3</sub>**  
 (Note: This is a polyatomic ion, meaning that it is a covalent compound with a charge. You will become experts on them in Unit 7.)
  
4. In your own words, define **polarity**:  
**One atom pulls more strongly on the atoms in a bond. Unequal distribution of electrons in a bond. The one Border collie pulls more strongly on the toy than the other Border collie, but instead of border collies they are atoms and instead of a chew toy it is electrons. So on and so forth.**
  
5. For each bond listed below, label the polarity by drawing an arrow in the correct direction (pointing towards the more electronegative atom).
  - a.  $\overset{\text{S}}{\text{---}} \overset{\text{O}}{\text{---}} \rightarrow$
  - b.  $\overset{\text{H}}{\text{---}} \overset{\text{Cl}}{\text{---}} \rightarrow$
  - c.  $\overset{\text{C}}{\text{---}} \overset{\text{F}}{\text{---}} \rightarrow$

# Chemistry Chapter 6 Answer Key File

**S Marginson**



**Chemistry Chapter 6 Answer Key File:**

## Reviewing **Chemistry Chapter 6 Answer Key File**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Chemistry Chapter 6 Answer Key File**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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