

Acids and Bases Webquest #2

(Types all answers in **RED**)

The degree of acidity or alkalinity (basic) is important in organisms. The body must consistently maintain a near neutral pH(7) in the blood and other tissues. In addition, the body produces buffers that can neutralize acids. Acids and bases conditions in the body occur due to different metabolic pathways producing varying levels throughout the body.

1. What does acidity mean?

A solution means low hydrogen ion concentration or a high concentration of hydroxide.

2. What pH scale represents acidity?

Acidity is measured from 0 to 7 on the pH scale. 7 is the neutral level of pH.

3. What characteristics of the acid constitute the balance of?

Hydrogen ions constitute the pH level. A high hydrogen ion concentration in the body is

4. What chemicals does the body produce to keep neutral pH?

5. Buffers _____ pH in the body.

6. Acids and base conditions occur due to _____ reactions in the body.

Water is one of the most important molecules in the body. Cells are made mostly of water and water is required for almost every metabolic reaction in the body. The forces of attraction between water molecules is so strong that the transportation of gas molecules can't easily remove the hydrogen from other water molecules. This reaction is known as dissociation, and it takes place in our cells. Water (H₂O) dissociates into H⁺ and OH⁻ ions. A hydrogen ion is a molecule in which the H⁺ ion is called the hydroxide ion, while the OH⁻ ion is called the hydroxide ion, while the H⁺ ion is called the hydrogen ion.

Acid And Base Webquest Answers

Emilie Sanchez



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