

1.1 Catalysts & activation energy

CATALYST - speeds up reaction

without affecting themselves/
the products

- allow for lower temps
- higher faster
- reduce activation energy
(same overall energy change)



ENZYMES - biological catalyst

- have active site, denatured
by incorrect conditions

GROUPS IN THE PERIODIC TABLE (SC11)

RATES OF REACTION (SC10)

HEAT ENERGY CHANGE IN CHEMICAL
REACTIONS (SC19)

1.2 energy changes in
reactions

- reaction happens when
particles collide enough
to generate activation
energy



- BONDS:
- breaking -
need energy
endothermic
 - making - releases
energy exothermic

1.3A exothermic & endothermic (temp, inc)

EXO - transfers energy INTO surroundings

ENDO - transfer energy FROM surroundings
(temp dec)

REACTION PROFILE - demonstrates energy
change

- neutralisation and displacement
ALWAYS ~~exo~~ exothermic

Chemistry Ch 2 answers Study Guide

J Elliott



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