Name: Amanda p7

Properties of Water Worksheet

1. The hydrogen and oxygen atoms are held together by Covlalent bonds.
2. The electrons are not shared equally creating a Polar molecule.
3. The polarity of water allows it to combine with most substances. Because of this, it is referred to as the Universal Solvent.
4. Water molecules stick to other water molecules. This property is called cohesion.
5. Hydrogen bonds form between adjacent water molecules because the positive charged hydrogen end of one water molecule attracts the negative charged oxygen end of another water molecule.
6. Water molecules stick to other materials due to its polar nature. This property is called cohesion.
7. Surface tension creates the skin-like surface formed due to the polar nature of water.
8. Water has a high specific heat, it can absorb a large amount of heat before a change in temperature takes place.
9. Evaporation is when water changes from a liquid to a gas.
10. Condensation is when water changes from a gas to a liquid.
11. Substances that are attracted to water are referred to as hydrophilic and are considered to be water loving.
12. Substances that are not attracted to water are referred to as hydrophobic and are considered to be water fearing.

Word Bank:

polar	hydrophilic	universal solvent	evaporation
cohesion	adhesion	combine	covalent
negative	high specific heat		surface tension
condensation	hydrophobic		positive

13. Draw and label a water molecule.

Identify the positive (+) and negative ends (-).

14. Give the chemical formula for water. H_2O

Assessing Water Quality Lab Answer Key

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Table of Contents Assessing Water Quality Lab Answer Key

1. Understanding the eBook Assessing Water Quality Lab Answer Key
    - The Rise of Digital Reading Assessing Water Quality Lab Answer Key
    - Advantages of eBooks Over Traditional Books
2. Identifying Assessing Water Quality Lab Answer Key
    - Exploring Different Genres
    - Considering Fiction vs. Non-Fiction
    - Determining Your Reading Goals
3. Choosing the Right eBook Platform
    - Popular eBook Platforms
    - Features to Look for in an Assessing Water Quality Lab Answer Key
    - User-Friendly Interface
4. Exploring eBook Recommendations from Assessing Water Quality Lab Answer Key
    - Personalized Recommendations
    - Assessing Water Quality Lab Answer Key User Reviews and Ratings
    - Assessing Water Quality Lab Answer Key and Bestseller Lists
5. Accessing Assessing Water Quality Lab Answer Key Free and Paid eBooks

- Assessing Water Quality Lab Answer Key Public Domain eBooks
- Assessing Water Quality Lab Answer Key eBook Subscription Services
- Assessing Water Quality Lab Answer Key Budget-Friendly Options

6. Navigating Assessing Water Quality Lab Answer Key eBook Formats
   - ePub, PDF, MOBI, and More
   - Assessing Water Quality Lab Answer Key Compatibility with Devices
   - Assessing Water Quality Lab Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
   - Adjustable Fonts and Text Sizes of Assessing Water Quality Lab Answer Key
   - Highlighting and Note-Taking Assessing Water Quality Lab Answer Key
   - Interactive Elements Assessing Water Quality Lab Answer Key
8. Staying Engaged with Assessing Water Quality Lab Answer Key
   - Joining Online Reading Communities
   - Participating in Virtual Book Clubs
   - Following Authors and Publishers Assessing Water Quality Lab Answer Key
9. Balancing eBooks and Physical Books Assessing Water Quality Lab Answer Key
   - Benefits of a Digital Library
   - Creating a Diverse Reading Collection Assessing Water Quality Lab Answer Key
10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
11. Cultivating a Reading Routine Assessing Water Quality Lab Answer Key
    - Setting Reading Goals Assessing Water Quality Lab Answer Key
    - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Assessing Water Quality Lab Answer Key
    - Fact-Checking eBook Content of Assessing Water Quality Lab Answer Key
    - Distinguishing Credible Sources
13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development

    - Exploring Educational eBooks
14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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