Name_____

Balancing Equations Worksheet – Answers

Note to students: It is acceptable to leave spaces blank when balancing equations – blank spaces are interpreted as containing the number "1".

1) $Na_3PO_4 + 3\ KOH \rightarrow 3\ NaOH + K_3PO_4$

2) $MgF_2 + Li_2CO_3 \rightarrow MgCO_3 + 2\ LiF$

3) $P_4 + 3\ O_2 \rightarrow 2\ P_2O_3$

4) $2\ RbNO_3 + BeF_2 \rightarrow Be(NO_3)_2 + 2\ RbF$

5) $2\ AgNO_3 + Cu \rightarrow Cu(NO_3)_2 + 2\ Ag$

6) $CF_4 + 2\ Br_2 \rightarrow CBr_4 + 2\ F_2$

7) $2\ HCN + CuSO_4 \rightarrow H_2SO_4 + Cu(CN)_2$

8) $GaF_3 + 3\ Cs \rightarrow 3\ CsF + Ga$

9) $BaS + PtF_2 \rightarrow BaF_2 + PtS$ (already balanced)

10) $N_2 + 3\ H_2 \rightarrow 2\ NH_3$

11) $2\ NaF + Br_2 \rightarrow 2\ NaBr + F_2$

12) $Pb(OH)_2 + 2\ HCl \rightarrow 2\ H_2O + PbCl_2$

13) $2\ AlBr_3 + 3\ K_2SO_4 \rightarrow 6\ KBr + Al_2(SO_4)_3$

14) $CH_4 + 2\ O_2 \rightarrow CO_2 + 2\ H_2O$

15) $2\ Na_3PO_4 + 3\ CaCl_2 \rightarrow 6\ NaCl + Ca_3(PO_4)_2$

16) $2\ K + Cl_2 \rightarrow 2\ KCl$

17) $2\ Al + 6\ HCl \rightarrow 3\ H_2 + 2\ AlCl_3$

18) $N_2 + 3\ F_2 \rightarrow 2\ NF_3$

19) $SO_2 + 2\ Li_2Se \rightarrow SSe_2 + 2\ Li_2O$

20) $2\ NH_3 + H_2SO_4 \rightarrow (NH_4)_2SO_4$

HW ✔ _____

HANDOUT

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