

Activity Based Questions

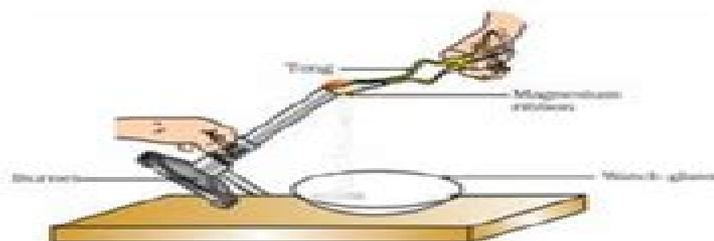
Class X (Science)

Chapter 1: Chemical Reactions and Equations

Activity 1.1: Burning of magnesium ribbon.

Observations:

- It is an example of chemical reaction between a metal (magnesium) and oxygen.
- In this reaction magnesium reacts with oxygen (present in the air).
- Magnesium ribbon burn with a **dazzling white flame**.
- In this reaction, it gives rise **magnesium oxide (MgO)** in the form of **white powder** (ash).
- It is an example of **combination reaction**.



- Before the start of the activity, the ribbon should be rub with a sandpaper.

Reason: For removing the oxide layer of magnesium. It reacts with oxygen slowly.

- Which change take place in this process?

Ans: Change of colour (white) & formation of a new substance (MgO).

Activity 1.2: Reaction of zinc and dilute sulphuric acid or hydrochloric acid.

Observations:

- In this activity zinc metal reacts with dilute sulphuric acid.
- Hydrogen gas evolves (releases) in this process.

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Lingjun Ying



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